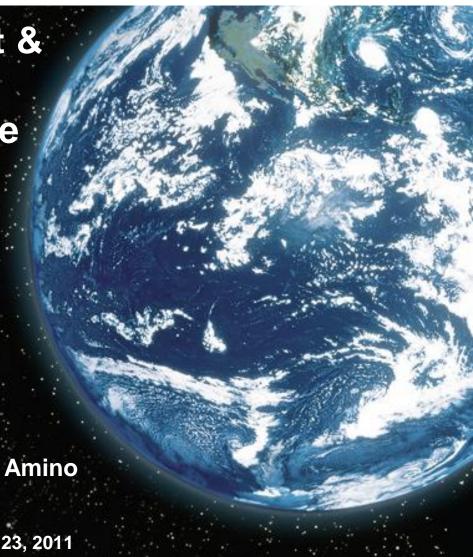
Slipstream Development & Testing of Post-Combustion CO₂ Capture and Separation Technology for Existing Coal-Fired Plants

DOE Contract DE-FE0003714
John L Winkler, Siemens

2.5 MWe Slipstream utilizing Siemens Amino Acid Salt (AAS) Solvent PostCap

NETL CO2 Capture Technology Meeting, August 23, 2011



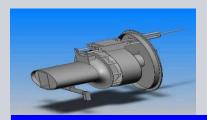
Siemens Energy, Inc. **Environmental Systems & Services Expanded Products for Global Markets**











- Carbon Capture Technology (CC)
 - Siemens PostCap Process
- Flue Gas Desulfurization (FGD)
 - Dry FGD
 - Mercury Control
 - Wet FGD
 - Wet ESP
- **Electrostatic Precipitators (ESP)**
 - **HaRDE**
 - **VIGR**
- Fabric Filters (FF)
 - Pulse Jet
 - Cartridge
 - Reverse Air
- NO_x and Ancillary Products
 - Low NO_X Burners
 - Overfire Air
 - SNCR/SCR
 - **Boiler Design Upgrades**











Project Overview – Project Participants

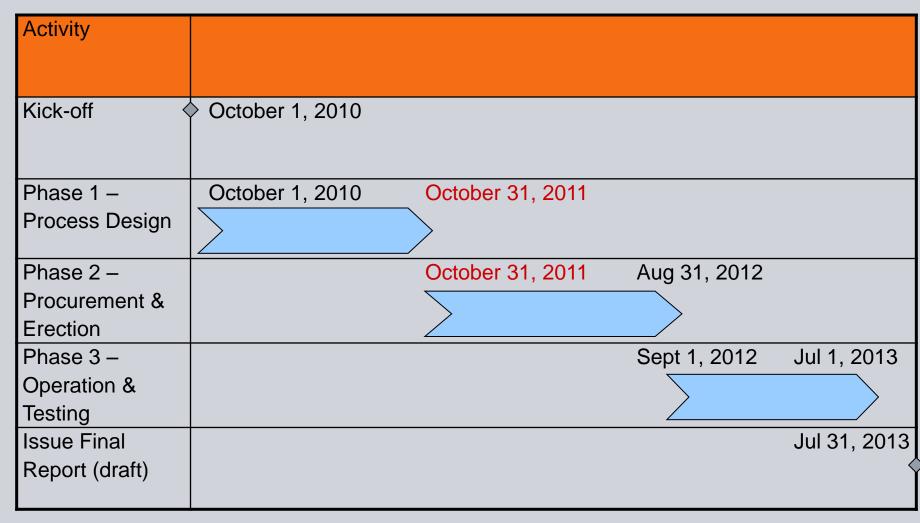
- DOE NETL Project Sponsor / Awarding Agency
- Siemens Energy, Inc. –
 Technology, Equipment, and Installation provider. Co-Sponsor
- Tampa Electric Company –
 Proposed Host Site Provider,
 Big Bend Plant Units 1&2



Project Objectives

- Demonstrate the ability of POSTCAP technology to achieve 90% CO₂ removal and approach a 35% increase of cost of electricity produced.
 - AAS technology <u>can</u> reach 90% CO2 removal (already proven)
 - Challenge of approaching 35% increase in COE
- Demonstrate the scalability and feasibility of progressing the POSTCAP technology to full-scale commercial application (550 MW) on post-combustion CO₂ capture for coal fired power plants and to full-scale commercial application for industrial sources of CO₂ emissions.
 - Proving scalability
 - Proving feasibility

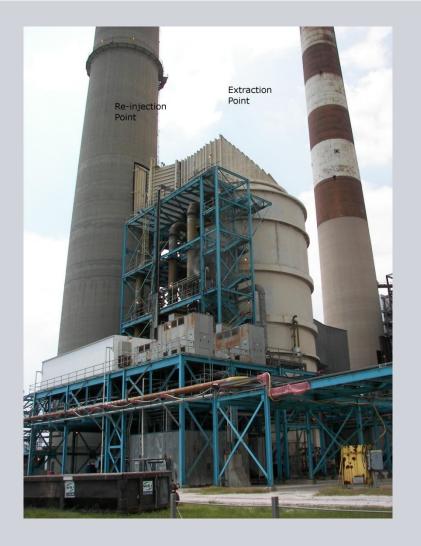
Project Overview – Overall Project Performance Dates **SIEMENS** Award, October 1, 2010



Project Scope

Slipstream Project Work

- Preliminary Engineering Phase1
- Detailed Engineering & Design (three month extension) - Phase1
- Development of Equipment/Procurement Packages (three month extension) - Phase1
- Retrofit installation into existing Big Bend scrubber outlet – Phase2
- Commissioning, Operation and Testing of plant – Phase3
- Data Analysis, Report out Phase3





Project Overview - Funding

	Budget Period 1		Budget Period 2		Budget Period 3	
	Process Design		Procurement & Erection		Operation & Testing	
	Oct 2010 – Oct 2011		Oct 2011 – Aug 2012		Sep 2012 – Oct 2013	
	DOE	Siemens	DOE	Siemens	DOE	Siemens
	share	share	share	share	share	share
Split (80/20) over 13 qtrs	\$1,411 K	\$353 K	\$10,580 K	\$2,645 K	\$3,008 K	\$752 K
DOE Project Share Total					\$15,000 K	
Siemens Project Share Total					\$ 3,750 K	

Status Report – Phase I, Detail Design and Engineering



Challenges:

- significant underground obstructions found in the area of the proposed plant
- problems in developing steam supply for desorption column
- resulting engineering delays and inefficiencies

Result:

- Three month extension granted to Phase 1 portion of project schedule by DOE
- Additional costs incurred by Siemens for engineering



Project Milestones

Activity ID	Activity Name	(*)Early Start	(*)Early Fin	Actual Finish
PHASE 1				
SE0900	Task 1.0 Project Management and Planning	10/1/2010	7/31/2011	10/31/2011
SE1012	Task 2.0 Generation of Heat & Material Balance	10/1/2010	11/30/201	
SE1013	Task 3.0 Generation of PFD's	10/1/2010	11/30/201	
SE1014	Task 4.0 Plant Interface Engineering	10/1/2010	6/30/2011	
SE1015	Task 5.0 Preliminary Design	10/1/2010	2/28/2011	
SE1045	Task 6.0 Detail Design	3/1/2011	7/31/2011	10/31/2011

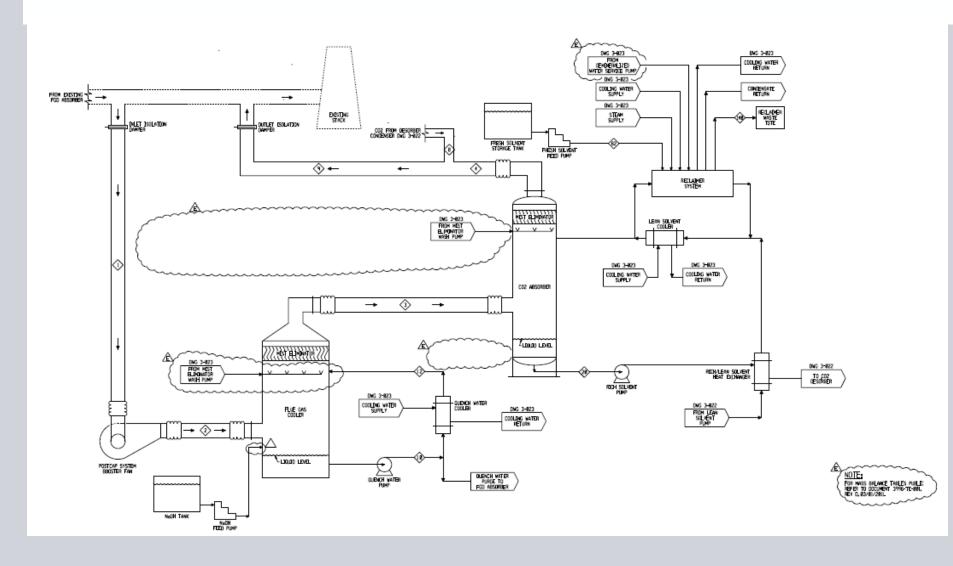


Project Methodology

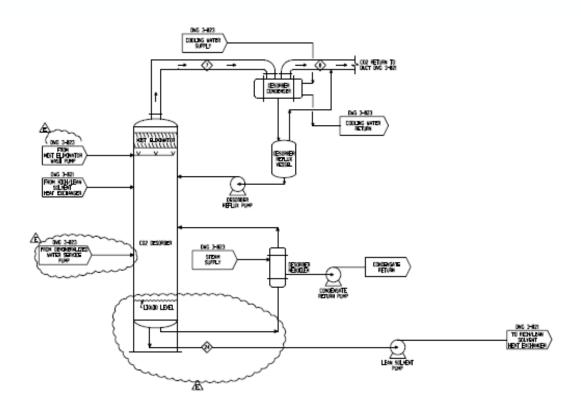
Schedule

Task 1.0 Preliminary Design Engineering	10/1/2010	2/28/2011
Task 2.0 Generation of Heat & Material Balance	10/1/2010	11/30/2011
Task 3.0 Generation of PFD's	10/1/2010	11/30/2011
Task 4.0 Plant Interface Engineering	10/1/2010	6/30/2011
Task 5.0 Preliminary Design	10/1/2010	2/28/2011
Task 6.0 Detail Design/Engineering	3/1/2011	10/31/2011
Task 7.0 Site Civil Work	8/1/2011	11/31/2011
Task 8.0 Plant Equipment & Material Procurement	8/1/2011	8/30/2012
Task 9.0 Complete Erection of Structural Steel	9/1/2011	2/15/2012
Task 10.0 Absorber/Desorber Installation Complete	9/1/2011	5/9/2012
Task 11.0 Mechanical Installation Complete	9/1/2011	5/9/2012
Task 12.0 Piping Component Installation	12/1/2011	8/30/2012
Task 13.0 Complete Electrical Installation	2/1/2012	7/13/2012
Task 14.0 Complete I&C Installation	2/1/2012	8/30/2012
Task 15.0 Test Plan Development	6/1/2012	8/31/2012
Task 16.0 Phase 3 - Commissioning, Testing, and Reporting	9/1/2012	7/31/2013
Task 17.0 Perform System Hydro Test	9/1/2012	10/1/2012
Task 18.0 Perform Plant Startup	10/31/2012	12/3/2012
Task 19.0 Steady State Plant Operation	12/3/2012	7/1/2013
Task 20.0 Perform Testing	12/3/2012	7/1/2013
Task 21.0 Data Analysis	7/2/2013	7/31/2013
Task 22.0 Decommission	8/1/2013	10/31/2013
Task 23.0 Disassemble	8/1/2013	10/31/2013

POST CAP PFD (1 of 2)



POST CAP PFD (2 of 2)



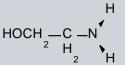


Technology Fundamentals – Amino Acid Salt is the basis of our solvent

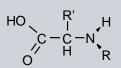




Ammonia

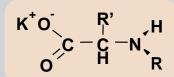


MEA

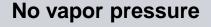


......

Amino acid



Amino acid salt



Chemically stable

Naturally present

Salts have no vapor pressure

- No thermodynamic solvent emissions
- Not flammable
- Not explosive
- Odorless
- No inhalation risk



Negative ion is less sensitive to O₂

Low degradation

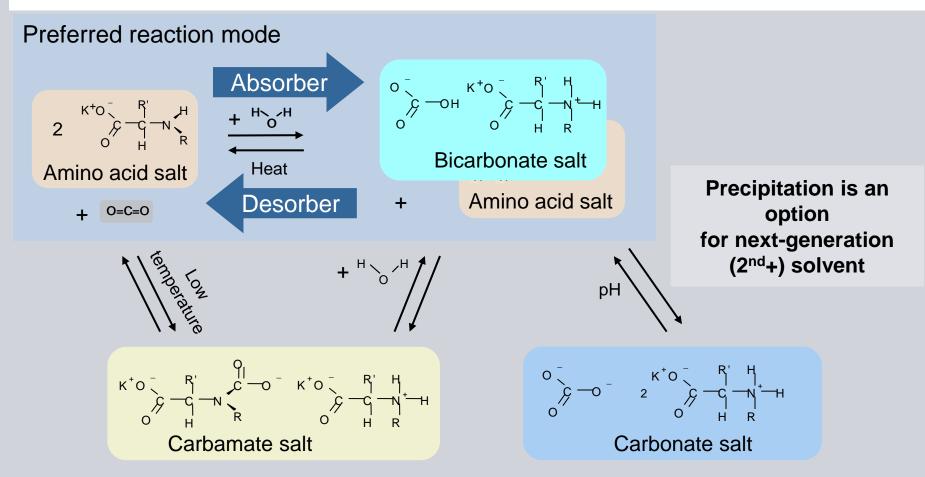
Amino acids are naturally present

- Biodegradable
- Nontoxic
- Environmentally friendly





Technology Fundamentals



Intermediate reaction products fulfill the same HSE standard as amino acid salts



Technology Background - Solvent Stability and Current Performance

Components	Amino Acid Salt w/o H₂O wash	
VOC	not detectable	
Formaldehyde	not detectable	
Methylamine	not detectable	
∑Nitrosamines	not detectable	
Ammonia	<1 ppm	

- the solvent is highly stable and does not lead to measurable loss of active substance due to degradation
- by-products in the liquid phase are salts with no vapor pressure
- No production of any mentionable amounts of emissions
- small amounts of heat stable salts (HSS) and nitrosamines will be removed with reclaimers

The amino-acid salt is stable against thermal stress and oxygen environments

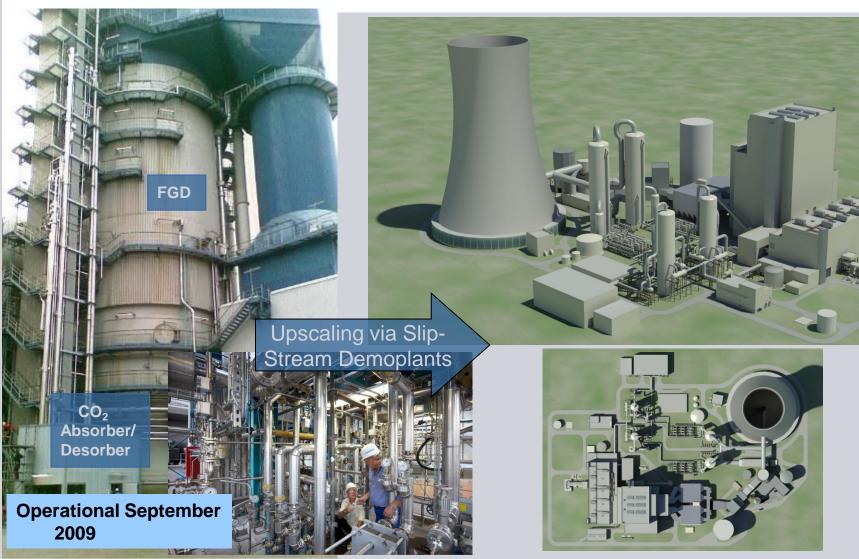
Background - Siemens lab plant for CO₂ capture tests at Frankfurt Hoechst Industrial Park

SIEMENS



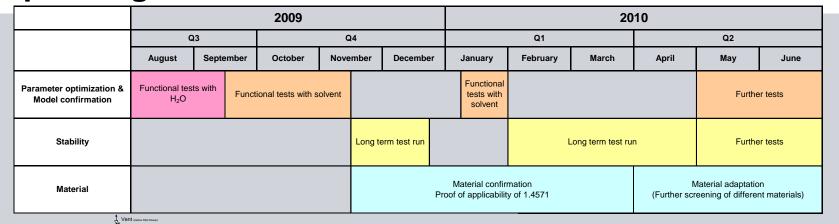
Background – E.ON Energie's Staudinger 100 kW Pilot Plant

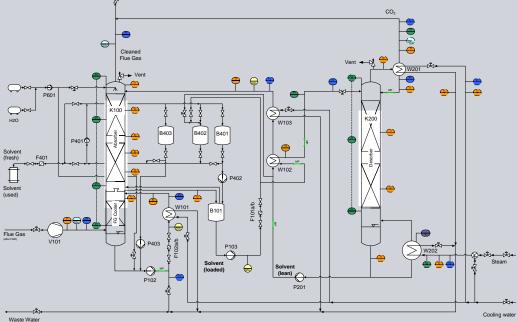




Staudinger Pilot Test Program – 4,500 operating hours as of June 2011







- Since the start up phase in August 2009 a large amount of test results have been collected
- Test results at different operating conditions were used to validate the existing process model
- Approx. 150 measurements were available to fit the model predictions to reality



Crystallization

Absorption capacity

- A high CO₂ loading of the solvent was achieved
- Various SO_x concentrations can be safely adjusted to investigate the behaviour of active AAS substance

Crystallization behavior for different solvent concentrations

- Crystallization behaviour as a function of temperature and CO₂ loading was extensively studied and is well understood
 - Several process arrangements were made in order to prevent crystallization

Crystallization effects were safely avoided

Solvent Stability



Results

Components	Amino Acid Salt w/o H₂O wash		
VOC	not detectable		
Formaldehyde	not detectable		
Methylamine	not detectable		
∑Nitrosamines	not detectable		
Ammonia	<1 ppm		

- the solvent is highly stable and does not lead to measurable loss of active substance due to degradation
- by-products in the liquid phase are salts with no vapor pressure
- No production of any mentionable amounts of emissions
- small amounts of heat stable salts (HSS) and nitrosamines will be removed with a reclaimer

The amino-acid salt is stable against thermal stress and oxygen environments!



Corrosion Study

- Several Test Coupons (ST 1.4571*)
 were investigated from TÜV Süd according DIN 50905
- No occurrence of local corrosion effects has been observed
- Similar qualifications have been conducted for lower grade materials (carbon steel 1.0425, st 1.4541, st 1.4525) with generally good results



*st 1.4571 = EN X6CrNiMoTi17-12-2 or AISI/SAE 316 Ti

Use of material stainless steel 1.4571 was confirmed, good potential for lower grade steels

Staudinger Pilot Plant Operation Results and Conclusions



Efficiency

Low energy consumption

- < 6% pts efficiency loss</p>
- 2.7 GJ heat consumption

High capture rate: >90 % proven

- EU prerequisites can be easily met
- Low CO₂ capture cost

Emissions

 No predictable solvent and nitrosamine salt emissions

- Add'l washing step avoided
- Permission less critical

Solvent stability

- < 1%/year solvent degradation by O₂
- High thermal stability
- Smart reclaimer concept for SO₂ applicable
- · Low refill requirements
- · Low reclaiming costs

Hardware

 Corrosion tests for construction material in all parts of capture plant

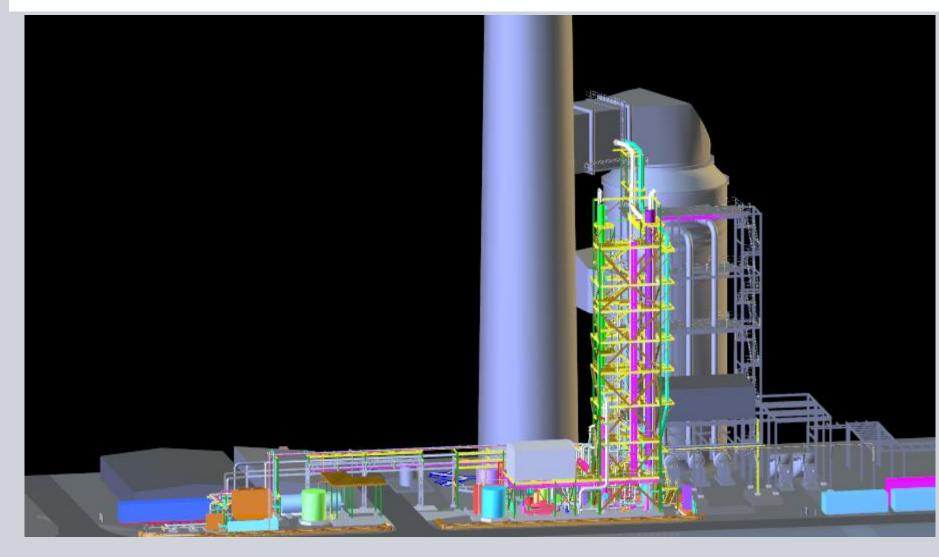
- Standard construction material applicable
- · Reduced investment costs



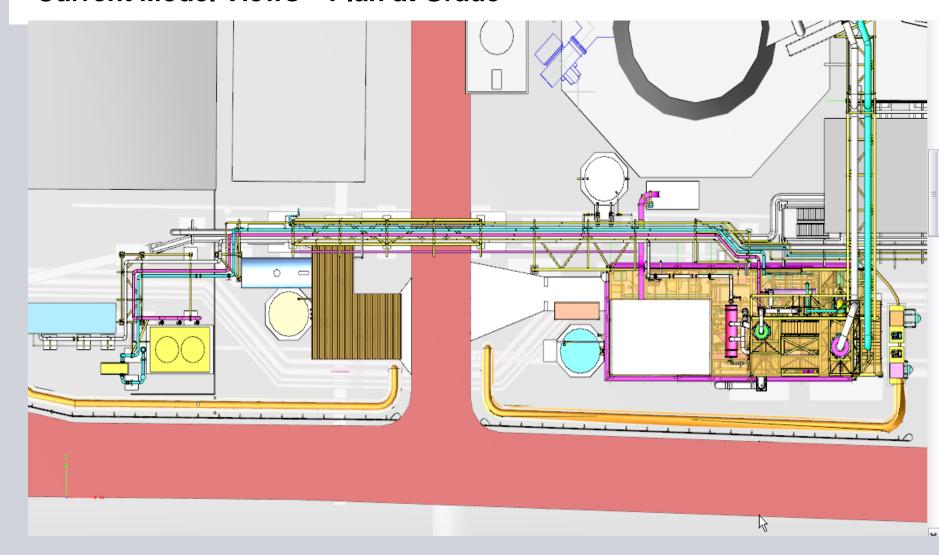
Current Model Views



Current Model Views

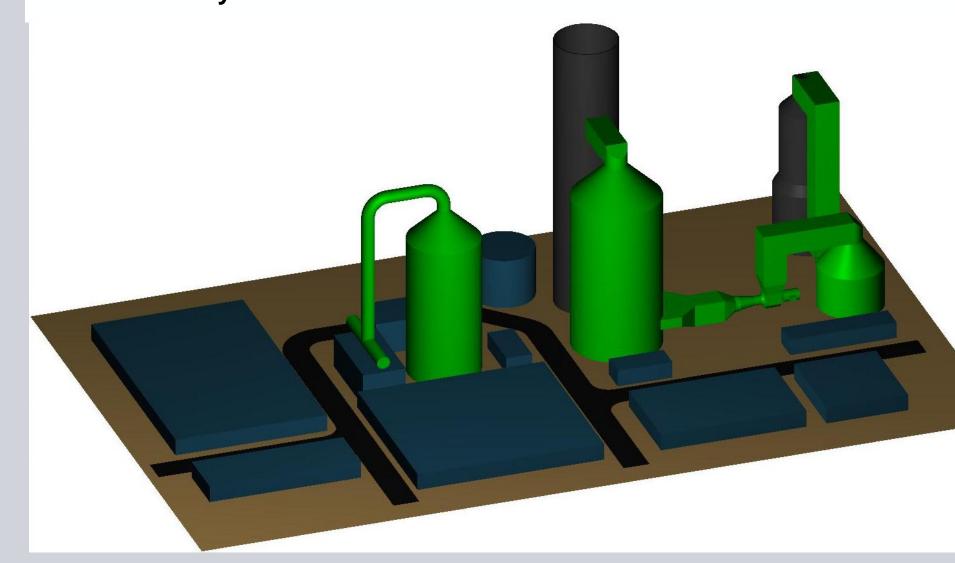


Current Model Views – Plan at Grade



Techno-economic Study of PostCap Application 550 MW Utility Plant Case - Model







Industrial Applications

- Cement Production
- Iron and Steel
- IGCC



Disclaimer

Disclaimer. This report was prepared as an account of work sponsored by an agency of the United States Government. Neither the United States Government nor any agency thereof, nor any of their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference here into any specific commercial product, process, or service by tradename, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.

Acknowledgement. This material is based up on work supported by the Department of Energy, National Energy Technology Laboratory under Award Number DE-FE0003714.